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Chemically-induced ultraweak photon emission (UPE) — Measurement as an analysis method of e. regri. degradation of polymeric material



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Foreword

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This document was prepared by Technical Committee ISO/TC 61, *Plastics*, Subcommittee SC 6, *Ageing, chemical and environmental resistance*.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Chemically-induced ultra-weak photon emission (UPE) — Measurement as an analysis method of degradation of polymeric material

1 Scope

This document provides a method for assessing the very early oxidative degradation state of polymers during outdoor weathering and indoor accelerated weathering tests and the influence of various additives can also be evaluated.

This method is based on an analytical method in which the degree of progress of the oxidative degradation reaction of a polymer is sensitively detected by measurement of chemically induced ultraweak photon emission (UPE).

2 Normative references

There are no normative references in this document.

3 Terms and definitions

For the purpose of this document, the following terms and definitions apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at https://www.iso.org/obp
- IEC Electropedia: available at http://www.electropedia.org

3.1

ultra-weak photon emission

phenomenon of light emission observed when electrons are excited to at least one-level higher energy level due to the energy produced by chemical reaction and then are deactivated from the higher energy levels to the ground state

Note 1 to entry: UPE is often used synonymously with "chemiluminescence". "Chemiluminescence" means luminescence accompanied with a chemical reaction.

3.2

dark current value

numeric data measured in a state where the detection element is not irradiated with light

3.3

background

numeric data measured in a state where samples are not placed in a sample container

3.4

oxidation induction time

OIT

time when the emission intensity rapidly increases due to the disturbance of the equilibrium between the generation rate and the disappearance rate of the peroxide produced in plastics

Note 1 to entry: The oxidation induction time varies according to the type of antioxidant, additive concentration, measurement temperature, ageing status, etc.