Jewellery and precious metals - Determination of platinum - Gravimetry using ammonium chloride (ISO 11210:2023)



EESTI STANDARDI EESSÕNA

NATIONAL FOREWORD

See Eesti standard EVS-EN ISO 11210:2023 sisaldab Euroopa standardi EN ISO 11210:2023 ingliskeelset teksti.

This Estonian standard EVS-EN ISO 11210:2023 consists of the English text of the European standard EN ISO 11210:2023.

Standard on jõustunud sellekohase teate avaldamisega EVS Teatajas.

This standard has been endorsed with a notification published in the official bulletin of the Estonian Centre for Standardisation and Accreditation.

Euroopa standardimisorganisatsioonid on teinud Euroopa standardi rahvuslikele liikmetele kättesaadavaks 15.03.2023.

Date of Availability of the European standard is 15.03.2023.

Standard on kättesaadav Eesti Standardimis-ja Akrediteerimiskeskusest.

The standard is available from the Estonian Centre for Standardisation and Accreditation.

Tagasisidet standardi sisu kohta on võimalik edastada, kasutades EVS-i veebilehel asuvat tagasiside vormi või saates e-kirja meiliaadressile <u>standardiosakond@evs.ee</u>.

ICS 39.060

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EUROPEAN STANDARD

EN ISO 11210

NORME EUROPÉENNE EUROPÄISCHE NORM

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Supersedes EN ISO 11210:2016

English Version

Jewellery and precious metals - Determination of platinum - Gravimetry using ammonium chloride (ISO 11210:2023)

Joaillerie, bijouterie et métaux précieux - Dosage du platine - Méthode gravimétrique utilisant le chlorure d'ammonium (ISO 11210:2023)

Schmuck und Edelmetalle - Bestimmung von Platin -Gravimetrie mittels Ammoniumchlorid (ISO 11210:2023)

This European Standard was approved by CEN on 19 February 2023.

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

CEN-CENELEC Management Centre: Rue de la Science 23, B-1040 Brussels

European foreword

This document (EN ISO 11210:2023) has been prepared by Technical Committee ISO/TC 174 "Jewellery and precious metals" in collaboration with CCMC.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by September 2023, and conflicting national standards shall be withdrawn at the latest by September 2023.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN shall not be held responsible for identifying any or all such patent rights.

This document supersedes EN ISO 11210:2016.

Any feedback and questions on this document should be directed to the users' national standards body/national committee. A complete listing of these bodies can be found on the CEN website.

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Endorsement notice

The text of ISO 11210:2023 has been approved by CEN as EN ISO 11210:2023 without any modification.

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Foreword

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The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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This document was prepared by the European Committee for Standardization (CEN) Technical Committee CEN/SSM21, *Precious metals* — *Applications in jewellery and associated products*, in collaboration with ISO Technical Committee TC 174, *Jewellery and precious metals*, in accordance with the agreement on technical cooperation between ISO and CEN (Vienna Agreement).

This third edition cancels and replaces the second edition (ISO 11210:2014), which has been technically revised.

The main changes are as follows:

- extension of the scope of application to all precious metal alloys beyond the jewellery sector;
- clarification of the fineness for which the test is suitable;
- addition of a specific preparation for samples containing a significant amount of silver in <u>Clause 8</u>;
- addition of an alternative method for the filtration in Clause 8:
- harmonization of method with ISO 11490.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Jewellery and precious metals — Determination of platinum — Gravimetry using ammonium chloride

1 Scope

This document specifies a gravimetric method for the determination of platinum on a material considered homogeneous. The platinum content of the sample lies preferably between 50 and 999 parts per thousand (%) by mass. Fineness above 999 % can be determined using a spectroscopy method by difference (e.g. ISO 15093).

This method is also intended to be used as one of the recommended methods for the determination of fineness in jewellery alloys covered by ISO 9202.

2 Normative references

There are no normative references in this document.

3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

ISO and IEC maintain terminology databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at https://www.iso.org/obp
- IEC Electropedia: available at https://www.electropedia.org/

3.1

platinum sponge

platinum obtained after calcination of the diammonium hexachloroplatinate precipitate

4 Principle

The sample is dissolved in aqua regia. Platinum is precipitated with ammonium chloride. The diammonium hexachloroplatinate precipitate is converted by ignition to metallic platinum which is weighed.

If present, silver is separated as silver chloride.

Co-precipitated alloying elements are tested in the re-dissolved platinum sponge and measured using, for example, an inductively coupled plasma optical emission spectrometer (ICP-OES), and a correction applied.

5 Reagents

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade and only distilled water or water of equivalent purity.

- **5.1 Hydrochloric acid (HCl)**, with a mass fraction of approximately 30 % to 37 % of HCl.
- **5.2 Diluted hydrochloric acid**, consisting of a mix of one volume of hydrochloric acid $(\underline{5.1})$ and one volume of water.