INTERNATIONAL STANDARD



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Water quality — Determination of dissolved anions by liquid chromatography of ions —

Part 4:

Determination of chlorate, chloride and chlorite in water with low contamination

Qualité de l'eau — Dosage des anions dissous par chromatographie des ions en phase liquide —

Partie 4: Dosage des ions chlorate, chlorure et chlorite dans des eaux faiblement contaminées



Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 10304-4 was prepared by Technical Committee ISO/TC 147, *Water quality*, Subcommittee SC 2, *Physical, chemical and biochemical methods*.

ISO 10304 consists of the following parts onder the general title *Water quality* — *Determination of dissolved anions by liquid chromatography of ions:*

- Part 1: Determination of fluoride, chloride, promide, nitrate, nitrite, orthophosphate and sulfate in water with low contamination

- Part 2: Determination of bromide, chloride, nitrate mitrite, orthophosphate and sulfate in waste water
- Part 3: Determination of chromate, iodide, sulfite, thiogyanate and thiosulfate
- Part 4: Determination of chlorate, chloride and chlorite in water with low contamination.

Annexes A and B of this part of ISO 10304 are for information of y.

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International Organization for Standardization Case postale 56 • CH-1211 Genève 20 • Switzerland Internet central@iso.ch X.400 c=ch; a=400net; p=iso; o=isocs; s=central

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Introduction

- Method of detection:

- Applicability of the method:

- Guaranteeing the analytical quality (9.3):

- Calibration (9.1):

The essential minimum requirements of an ion chromatographic system applied within the scope of this part of ISO 10304 are the following:

- Resolution power of the column:

For the anion to be determined it is essential that the peak resolution does not fall below R = 1.3(clause 7, figure 3)

a) Measurement of the electrical conductivity with or without suppressor device

b) Spectrometric measurement (UV/VIS), directly or indirectly

c) Amperometric direct detection

Working ranges according to table 1

Calibration and determination of the linear working range (see ISO 8466-1). Use of the method of standard addition to special cases of application (9.2).

Vanidity check of the calibration function. Replicate determinations, if necessary.

The diversity of the appropriate and suitable assemblies and the procedural steps depending on them permit a general description only. A CAN RELATED DY THE

For further information on the analytical technique see referend

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Water quality — Determination of dissolved anions by liquid chromatography of ions —

Part 4:

Determination of chlorate, chloride and chlorite in water with low contamination

1 Scope

1:5 OC This part of ISO 10304 specifies a method for the determination of the dissolved anions chlorate, chloride, and chlorite in water with low contamination (e.g. drinking water, raw water or swimming pool water).

An appropriate pretreatment of the sample (e.g. dilution) and the use of a conductivity detector (CD), UV detector (UV) or amperometric detector (AD) make the working ranges given in table 1 feasible.

mg/I	
0.0015-40	
0,03 to 10	CD
0,1 to 50	CD
0,05 to 1	CD
0,1 to 1	UV; λ=207 nm to 220 nm
0,01 to 1	AD; 0,4 to 1,0 V
ted by the ion-exchange capacity of the o	columns. Dilute the sample in to the working range, it
ge for chlorite of 0,05 mg/l was obtained that it is difficult to obtain this with suffic this method.	d using calibration crecks, but the round robin trials cient accuracy. Thus great care shall be taken when
1	0,1 to 50 0,05 to 1 0,1 to 1 0,01 to 1 ted by the ion-exchange capacity of the or ge for chlorite of 0,05 mg/l was obtained that it is difficult to obtain this with suffi- this method.

Table 1 — Working ranges of the analytical method

The following standards contain provisions which, through reference in this text, constitute provisions of this part of ISO 10304. At the time of publication, the editions indicated were valid. All standards are subject to revision, and parties to agreements based on this part of ISO 10304 are encouraged to investigate the possibility of applying the most recent editions of the standards indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 5667-1:1980 Water quality - Sampling - Part 1: Guidance on the design of sampling programmes.

ISO 5667-2:1991 Water quality - Sampling - Part 2: Guidance on sampling techniques.

ISO 10304-4:1997(E)

- ISO 5667-3:1994 Water quality - Sampling - Part 3: Guidance on the preservation and handling of samples.
- ISO 8466-1:1990 Water quality - Calibration and evaluation of analytical methods and estimation of performance characteristics - Part 1: Statistical evaluation of the linear calibration function.
- ISO 10304-1:1992 Water quality - Determination of dissolved anions by liquid chromatography of ions -Part 1: Determination of fluoride, chloride, bromide, nitrate, nitrite, orthophosphate and sulfate in water with low contamination
- Water quality Determination of dissolved anions by liquid chromatography of ions -ISO 10304-2:1995 Par 2: Determination of bromide, chloride, nitrate, nitrite, orthophosphate and sulfate in waste water
- ISO 10304-3:1997 Water Water Water Water Water of dissolved anions by liquid chromatography of ions -Part 3: Determination of chromate, iodide, sulfite, thiocyanate and thiosulfate
- ISO 10530: 1992 Water quality Determination of dissolved sulfide - Photometric method using methylene blue Nont is at

3 Interferences

disinfection byproducts (e.g. chloroacetic acid) can 3.1 Organic acids such as mono- and dicarboxylic acids interfere.

3.2 Dissolved organics can react with the working electrode of the amperometric detector, causing a decrease in sensitivity.

3.3 The presence of fluoride, carbonate, nitrite and nitrate can care interference with the determination of chlorate, chloride and chlorite. The respective concentrations given in table 2 are typical for conductivity, UV and amperometric detectors.

3.4 Elevated loads of chloride and bromide can cause interference with the dependent of chlorite and chlorate. Remove chloride and bromide with the aid of special exchangers (8.2).

3.5 Solid particles and organic compounds (such as mineral oils, detergents, and hemic acids) shorten the life-time of the separator column. They are therefore eliminated from the sample prior to analysis clause 8).