



EESTI STANDARDI EESSÕNA NATIONAL FOREWORD

Käesolev Eesti standard EVS-EN ISO 14912:2006 sisaldab Euroopa standardi EN ISO 14912:2006 ingliskeelset teksti.	This Estonian standard EVS-EN ISO 14912:2006 consists of the English text of the European standard EN ISO 14912:2006.
Käesolev dokument on jõustatud 20.09.2006 ja selle kohta on avaldatud teade Eesti standardiorganisatsiooni ametlikus väljaandes.	This document is endorsed on 20.09.2006 with the notification being published in the official publication of the Estonian national standardisation organisation.
Standard on kättesaadav Eesti standardiorganisatsioonist.	The standard is available from Estonian standardisation organisation.
Käsitlusala: This International Standard defines the following quantities commonly used to express the composition of gas mixtures: - mole fraction, - mass fraction, - volume fraction, and - mole concentration, - mass concentration, - volume concentration.	mole fraction, - mass fraction, - volume fraction, and - mole concentration, - mass concentration, - volume concentration.
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<b>ICS</b> 71.040.40	
Võtmesõnad:	°O

Eesti Standardikeskusele kuulub standardite reprodutseerimis- ja levitamisõigus

## EUROPEAN STANDARD

## EN ISO 14912

## NORME EUROPÉENNE

**EUROPÄISCHE NORM** 

August 2006

71.040.40 **English Version** analysis - Conversion of gas mixture composition data (ISO 14912:2003) Analyse des gaz - Conversion des données de composition Gasanalyse - Umrechnung von de mélanges gazeux (ISO 14912:2003) Zusammensetzungsangaben für Gasgemische (ISO 14912:2003) This European Standard was approved by CEN on 21 July 2006. CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration. Up-to-date lists and bibliographical references concerning such national standards may be obtained on application to the Central Secretariat or to any CEN member. This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the Central Secretariat has the same status as the official versions. CEN members are the national standards bodies of Austria, Belgium, Cyprus, Czech Republic, Denmark, Estonia, Finland, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugal, Romania, Slovakia, Slovenia, Spain, Sweden, Switzerland and United Kingdom. Ochorado of the officer EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG Management Centre: rue de Stassart, 36 B-1050 Brussels

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## Foreword

The text of ISO 14912:2003 has been prepared by Technical Committee ISO/TC 158 "Analysis of gases" of the International Organization for Standardization (ISO) and has been taken over as EN ISO 14912:2006 by Technical Committee CEN/SS N21 "Gaseous fuels and combustible gas", the secretariat of which is held by CMC.

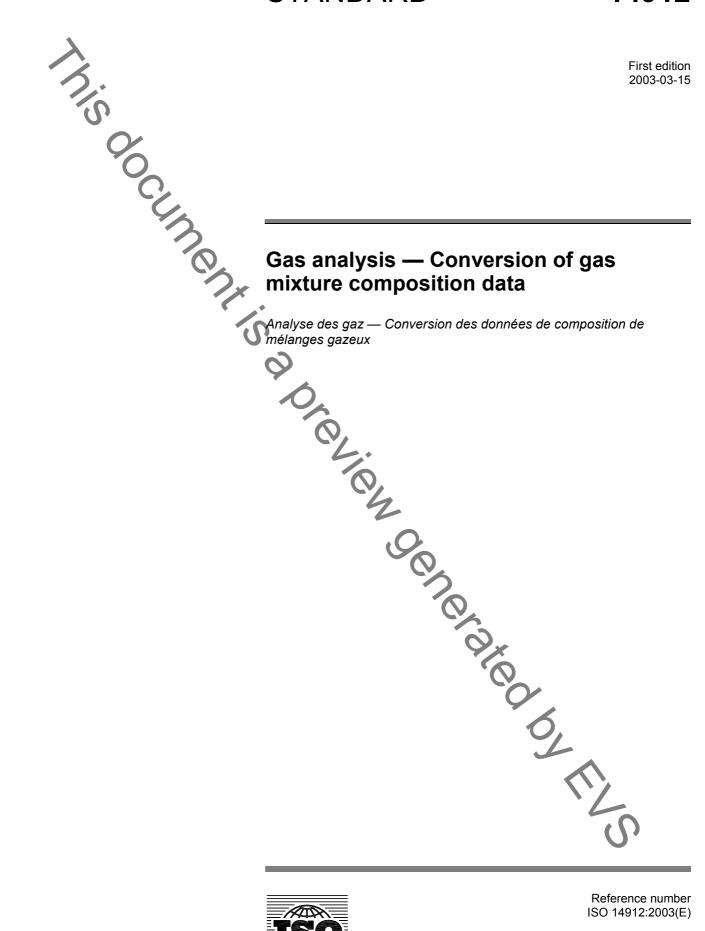
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Reference number ISO 14912:2003(E)

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Foreword

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International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

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ISO 14912 was prepared by Technical Committee ISO/TC 158, Analysis of gases.

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## Introduction

The objective of gas analysis is to determine the composition of gas mixtures. Gas mixture composition is expressed qualitatively in terms of specified mixture components of interest, called analytes, and the complementary gas. Gas mixture composition is expressed quantitatively by specifying the amount of every analyte in the mixture and the composition of the complementary gas.

For the purpose of specifying the amount of an analyte in a gas mixture, different quantities are in use. This diversity is due to the fact that in different applications different quantities have decisive advantages. Therefore procedures for conversion between different quantities are required.

In cases where these quantities involve the volumes of the analytes or the gas mixture or both, they depend on the state conditions, i.e. pressure and temperature, of the gas mixture. For each of these quantities, procedures for conversion between different state conditions are required.

As a crude approximation, all of the conversions referred to above can be performed on the basis of the Ideal Gas Law. In most cases, however, an accurate conversion has to take into account the real gas volumetric behaviour of the analyte and of the gas mixture. In particular, many conversions require values of the compression factor (or of the density) of the gas mixture.

This International Standard provides formally exact conversion procedures, based on fundamental principles, which fully account for real gas behaviour of pure gases and gas mixtures. In addition to these, approximate procedures for practical applications are described, designed for different levels of accuracy and available data. These approximations are necessary because measured gas mixture compression factors (or densities) are rarely available and therefore have to be estimated from component data. Uncertainty estimates are given which result from combining approximations in the conversion procedures with the uncertainties of the input data. Where conversions require real-gas volumetric data of pure gases or gas mixtures, these are expressed by compression factors. As equivalents, density data could be converted into compression factor data.

# analysis — Conversion of gas mixture composition data

This International Standard defines the following quantities commonly used to express the composition of gas mixtures:

- mole fraction,
- mass fraction.
- volume fraction,

and

- mole concentration,
- mass concentration,
- volume concentration.

For these quantities of composition, this International Standard provides methods for

- conversion between different quantities, and
- conversion between different state conditions.

Conversion between different quantities means calculating the numerical value of an analyte content in terms of one of the quantities listed above from the numerical value of the same analyte content, at the same pressure and temperature of the gas mixture, given in terms of another of these quantities. Conversion between different state conditions means calculating the numerical value of an analyte content, in terms of one of the quantities listed above, under one set of state conditions from the numerical value of the same quantity under another set of state conditions, i.e. pressure and temperature, of the gas mixture. Gas mixture composition can be converted simultaneously between different quantities of composition and different state conditions by combination of the two types of conversion.

This International Standard is applicable only to homogeneous and stable gas mixtures. Therefore any state conditions (pressure and temperature) considered need to be well outside the condensation region of the gas S T mixture and that of each of the specified analytes (see Annex A).

### Terms and definitions 2

For the purpose of this document, the following terms and definitions apply.

NOTE See also References [1] and [2] in the Bibliography.

### 2.1 Quantities for the expression of gas mixture composition

NOTE Further information concerning the terms defined in this subclause is given in 4.1.