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No Indoor air - Part 26: Sampling strategy for carbon dioxide (CO2) (ISO 16000-26:2012)



#### **EESTI STANDARDI EESSÕNA**

#### **NATIONAL FOREWORD**

| See Eesti standard EVS-EN ISO 16000-26:2012  | This Estonian standard EVS-EN ISO 16000-26:2012                                |  |
|--|--|--|
| sisaldab Euroopa standardi EN ISO 16000-26:2012  | consists of the English text of the European standard                          |  |
| ingliskeelset teksti.  | EN ISO 16000-26:2012.  |  |
| To the second se |  |  |
| ,  | This standard has been endorsed with a notification                            |  |
| avaldamisega EVS Teatajas.   | published in the official bulletin of the Estonian Centre for Standardisation. |  |
| Euroopa standardimisorganisatsioonid on teinud   | Date of Availability of the European standard is                               |  |
| ,  | 01.08.2012.  |  |
| kättesaadavaks 01.08.2012.   | 01.00.2012.  |  |
| nationalation of 1.00.2012.  |  |  |
| Standard on kättesaadav Eesti Standardikeskusest.  | The standard is available from the Estonian Centre for                         |  |
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ICS 13.040.20

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### **EUROPEAN STANDARD**

#### **EN ISO 16000-26**

## NORME EUROPÉENNE EUROPÄISCHE NORM

August 2012

ICS 13.040.20

#### **English Version**

# Indoor air - Part 26: Sampling strategy for carbon dioxide (CO<sub>2</sub>) (ISO 16000-26:2012)

Air intérieur - Partie 26: Stratégie d'échantillonnage du dioxyde de carbone (CO<sub>2</sub>) (ISO 16000-26:2012)

Innenraumluftverunreinigungen - Teil 26: Probenahmestrategie für Kohlendioxid (CO<sub>2</sub>) (ISO 16000-26:2012)

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

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#### **Foreword**

This document (EN ISO 16000-26:2012) has been prepared by Technical Committee ISO/TC 146 "Air quality" in collaboration with Technical Committee CEN/TC 264 "Air quality" the secretariat of which is held by DIN.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by February 2013, and conflicting national standards shall be withdrawn at the latest by February 2013.

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#### **Endorsement notice**

The text of ISO 16000-26:2012 has been approved by CEN as a EN ISO 16000-26:2012 without any modification.

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#### Introduction

In ISO 16000-1, general requirements relating to the measurement of indoor air pollutants and the important conditions to be observed before or during the sampling of individual pollutants or groups of pollutants are described.

This part of ISO 16000 describes basic aspects to be considered when working out a sampling strategy for the measurements of carbon dioxide in indoor air. It is intended to be a link between ISO 16000-1 and the analytical procedures.

This part of ISO 16000 presupposes knowledge of ISO 16000-1.

This part of ISO 16000 uses the definition for indoor environment defined in ISO 16000-1 and Reference [12] as dwellings having living rooms, bedrooms, DIY (do-it-yourself) rooms, recreation rooms and cellars, kitchens and bathrooms; workrooms or work places in buildings which are not subject to health and safety inspections with regard to air pollutants (for example, offices, sales premises); public buildings (for example hospitals, schools, kindergartens, sports halls, libraries, restaurants and bars, theatres, cinemas and other and in this s function rooms), and also cabins of vehicles and public transport.

The sampling strategy procedure described in this part of ISO 16000 is based on VDI 4300 Part 9.[11]

#### Indoor air —

#### Part 26:

## Sampling strategy for carbon dioxide (CO<sub>2</sub>)

#### 1 Scope

This part of ISO 16000 specifies the planning of carbon dioxide indoor pollution measurements. In the case of indoor air measurements, the careful planning of sampling and the entire measurement strategy are of particular significance since the result of the measurement can have far-reaching consequences, for example, with regard to ascertaining the need for remedial action or the success of such an action.

An inappropriate measurement strategy can lead to misrepresentation of the true conditions or, worse, to erroneous results.

This part of ISO 16000 is not applicable to the measurement strategy for carbon monoxide (CO).

NOTE See 5.1.

#### 2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 16000-1:2004, Indoor air — Part 1: General aspects of sampling strategy

#### 3 Properties, origin and occurrence of carbon dioxide

Carbon dioxide (CO<sub>2</sub>, CAS No. 124-38-9) is a natural constituent of atmospheric air, where it is present on average at a content of a little over 0.03 % (volume fraction; equal to about 600 mg/m³). CO<sub>2</sub> content measured in air is usually reported in the unit parts per million (1 ppm as a volume fraction is 1  $\mu$ mol/mol), 0.03 % as volume fraction being equivalent to 300 ppm. Some CO<sub>2</sub> reacts under indoor conditions with atmospheric humidity to form carbonic acid. CO<sub>2</sub> is colourless, odourless and without taste, readily water soluble and chemically stable under standard conditions. The CO<sub>2</sub> molecule can absorb part of the infrared radiation reflected by the Earth's surface as heat radiation and thus contributes to a process termed "greenhouse effect", which causes global warming.

Carbon dioxide plays a central role in the processes of life on Earth. As a result of plant activity (from carbon dioxide and water, under the action of sunlight in the presence of chlorophyll as catalyst), organic compounds (predominantly carbohydrates) are formed, as well as the oxygen which is essential for life on Earth. More or less in reverse to this process,  $CO_2$  is formed in the air as one of the end-products (in addition to water) of combustion of hydrocarbons. This process proceeds firstly in every type of combustion apparatus and fireplace, but secondly also plays an important role in the metabolism of living organisms. The  $CO_2$  formed in metabolic processes is released to the ambient air.

In the case of humans, the amount released depends on the extent of physical activity. For adult persons, the orders of magnitude listed in Table A.1 of the volume of CO<sub>2</sub> released can be assumed (see Annex A for more detailed explanations).

Since the beginning of industrialization, the  $CO_2$  concentration of the ambient air has been continuously rising.  $CO_2$  measurement sites in the past were usually linked with meteorological stations and were situated in clean air regions. One of the best-known  $CO_2$  measurement sites is on Mauna Loa in Hawaii. There, uninfluenced by any local  $CO_2$  source, the  $CO_2$  concentration increased, for example, from 316,0 ppm in 1959 to 369,4 ppm in

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