Juustu- ja sulatatud juustutooted. Kloriidisisalduse määramine. Potentsiomeetriline tiitrimismeetod

Cheese and processed cheese products - Determination of chloride content - Potentiometric titration method



FESTI STANDARDI FESSÕNA

NATIONAL FOREWORD

Käesolev Eesti standard EVS-EN ISO 5943 V2:2006 sisaldab Euroopa standardi EN ISO 5943:2006 ingliskeelset teksti.

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This Estonian standard EVS-EN ISO 5943 V2:2006 consists of the English text of the European standard EN ISO 5943:2006.

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ICS 67.100.30

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EUROPEAN STANDARD

EN ISO 5943

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English Version

Pheese and processed cheese products - Determination of formula content - Potentiometric titration method (ISO 5943:2006)

Fromages et fromages fondus - Détermination de la teneur en chlorures - Méthode par titrage potentiométrique (ISO 5943:2006) Käse und Schmelzkäse - Bestimmung des Chloridgehaltes - Potentiometrisches Titrationsverfahren (ISO 5943:2006)

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Foreword

This document (EN ISO 5943:2006) has been prepared by Technical Committee ISO/TC 34 "Agricultural food products" in collaboration with Technical Committee CEN/TC 302 "Milk and milk products - Methods of sampling and analysis", the secretariat of which is held by NEN.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by April 2007, and conflicting national standards shall be withdrawn at the latest by April 2007.

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> **IDF** 88

Fourth edition 2006-10-01

Cheese and processes products — Determination content — Potentiometric titration method

Promages et fromages fondus — Détermination de la teneur en chlorures — Méthode par titrage potentiométrique



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Foreword

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Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 5943 IDF 88 was prepared by Technical committee ISO/TC 34, *Food products*, Subcommittee SC 5, *Milk and milk products*, and the International Daio Federation (IDF). It is being published jointly by ISO and IDF.

This edition of ISO 5943 | IDF 88 cancels and replaces to 5943 | IDF 88:2004, of which it constitutes a minor revision. There was an error in the calculation given in 9.14 factor of 1 000 has been added and molar mass is now used.

Foreword

IDF (the International Dairy Federation) is a worldwide federation of the dairy sector with a National Committee in every member country. Every National Committee has the right to be represented on the IDF Standing Committees carrying out the technical work. IDF collaborates with ISO in the development of standard methods of analysis and sampling for milk and milk products.

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ISO 5943 IDF 88 was prepared by the International Dairy Federation (IDF) and Technical Committee ISO/TC 34, Food products, Subcommittee SC 5, Milk and milk products. It is being published jointly by IDF and ISO.

All work was carried out by the former wint ISO/IDF/AOAC Group of Experts, *Nitrate, nitrite and phosphorus in cheese*, under the aegis of its chairman or G. Bråthen (NO).

This edition of ISO 5943 IDF 88 cancels and replaces ISO 5943 IDF 88:2004, of which it constitutes a minor revision. There was an error in the calculation given in 9.1. A factor of 1 000 has been added and molar mass is now used.

Cheese and processed cheese products — Determination of chloride content — Potentiometric titration method

Scope

This International Standard specifies a potentiometric titration method for the determination of the chloride content of cheese and processed cheese products.

The method is applicable to all cheeses and processed cheese products containing more than 0,2 % (mass fraction) of chloride ion.

Terms and definitions

For the purposes of this document, the following terms and definitions apply.

chloride content of cheese and processed cheese products
mass fraction of substances determined by the projecture specified in this International Standard

It is expressed as a mass fraction, in percent, phloride ion or sodium chloride or any other chloride. NOTE

Principle

ded with nitric acid then the chloride ions are A test portion is suspended in water. The suspension is ac titrated potentiometrically with a silver nitrate standard solution.

Reagents

Use only reagents of recognized analytical grade, unless otherwise specified, and distilled or demineralized water or water of at least equivalent purity.

Silver nitrate standard solution, $c(AgNO_3) = 0.08 \text{ mol/l}$ to 0,12 mol/l. 4.1

Dissolve 13,6 g to 20,4 g of silver nitrate in water which is practically free from carbon dioxide and dilute to 1 000 ml. Standardize the solution against sodium chloride (NaCl), which has previously been dried at 300 °C, expressing the concentration of the silver nitrate standard solution to four decimal places.

Store the solution away from direct light.

Nitric acid, $c(HNO_3) \approx 4 \text{ mol/l.}$