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**Nuclear energy — Determination of  
nitrogen content in  $\text{UO}_2$ ,  $(\text{U,Gd})\text{O}_2$  and  
 $(\text{U,Pu})\text{O}_2$  sintered pellets — Inert gas  
extraction and conductivity detection  
method**

*Énergie nucléaire — Dosage de la teneur en azote des pastilles frittées  
d' $\text{UO}_2$ ,  $(\text{U,Gd})\text{O}_2$  et  $(\text{U,Pu})\text{O}_2$  — Méthode d'extraction par gaz inerte et  
méthode de mesurage de la conductivité*



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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see [www.iso.org/directives](http://www.iso.org/directives)).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see [www.iso.org/patents](http://www.iso.org/patents)).

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For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: [Foreword - Supplementary information](#)

The committee responsible for this document is ISO/TC 85, *Nuclear Energy*, Subcommittee SC 5, *Fuel Technology*.

# Nuclear energy — Determination of nitrogen content in $\text{UO}_2$ , $(\text{U,Gd})\text{O}_2$ and $(\text{U,Pu})\text{O}_2$ sintered pellets — Inert gas extraction and conductivity detection method

## 1 Scope

This International Standard describes a procedure for measuring the nitrogen content of  $\text{UO}_2$ ,  $(\text{U,Gd})\text{O}_2$ , and  $(\text{U,Pu})\text{O}_2$  pellets. Nitrogen in nuclear fuel may be present either as elemental nitrogen or chemically combined in the form of nitrogenous compounds. The technique described herein serves to determine the total content of nitrogen excluding those compounds whose decomposition temperature is above 2 200 °C (most notably Pu and U nitrides).

## 2 Principle

For determining the nitrogen content of  $\text{UO}_2$ ,  $(\text{U,Gd})\text{O}_2$ , or  $(\text{U,Pu})\text{O}_2$  pellets, an analyser is employed which operates according to the carrier gas principle, using helium as the carrier gas, the nitrogen content being determined in a thermal conductivity cell.

The weighed samples to be analysed are heated in a degassed high purity graphite crucible at a temperature of more than 1 770 °C in a helium atmosphere. This high temperature destroys the majority of the nitrogen bearing compounds and causes nitrogen to be released along with other gaseous components like CO,  $\text{CO}_2$ , and  $\text{H}_2$ . The released gases are then swept by the carrier gas through oxidation and absorption columns to trap interfering species. The nitrogen passes through without being retained, and its quantity is subsequently measured in the thermal conductivity cell.

## 3 Interferences

The samples will not be heated to temperatures above 2 200 °C since this would cause a reaction to take place between the  $\text{UO}_2$ ,  $(\text{U,Pu})\text{O}_2$ , or  $(\text{U,Gd})\text{O}_2$  pellet and the graphite, resulting in large quantities of  $\text{CO}_2$  gas being released. These large quantities of gas can be not completely trapped and would lead to errors in conductivity measurement.

## 4 Reagents and materials

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade. The reagents and materials below serve as examples to be used according to manufacturer's recommendation.

### 4.1 Helium.

Use helium as carrier gas with a purity of a volume fraction  $\geq 99,995\%$ .

### 4.2 Nitrogen.

If nitrogen is used as calibration gas it will be of a purity  $\geq 99,998\%$  in volume fraction.