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Nuclear fuel technology — Controlledpotential coulometric assay of plutonium

nolog lométrie. Technologie du combustible nucléaire — Dosage du plutonium par





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Contents			Page
Fore	word		iv
1	Scope	2	1
2	Norm	native references	1
3	50	s and definitions	
4		iple	
5	_	ents	
6		ratus	
7	7.1 7.2	Plutonium determination 7.1.1 Weighing the test sample, with an uncertainty of ±0,01 %, K = 1 7.1.2 Preparation of the test sample. 7.1.3 Electrode pre-treatment. 7.1.4 Electrical calibration of the current integration system. 7.1.5 Formal potential determination 7.1.6 Coulometric blank determination. 7.1.7 Plutonium measurement. Analysis of subsequent test samples.	
8	8.1 8.2 8.3 8.4 8.5	Calculation of the electrical calibration factor Calculation of the blank Fraction of electrolysed plutonium Plutonium content Quality control	14 15 16
9	Chara 9.1 9.2 9.3	Repeatability Confidence interval Analysis time	16 17
10	Inter	ferences	17
11	Proce 11.1 11.2 11.3 11.4 11.5 11.6 11.7 11.8 11.9	Accountability measurements and reference material preparation Process control measurements Measurement cell design Electrolyte and electrode options Test sample size Background current corrections Correction for iron Control-potential adjustment Calibration methodologies	21 21 22 22 22 23 24
		rmative) Purification by anion-exchange separation	
Ann	ex B (no	rmative) Determination of formal potential, E ₀	27
Rihl	iogranh	V	28

Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see the following URL: www.iso.org/iso/foreword.html.

The committee responsible for this document is Technical Committee ISO/TC 85, *Nuclear energy, nuclear technologies, and radiological protection,* Subcommittee SC 5, *Nuclear fuel cycle*.

This third edition cancels and replaces the second edition (ISO 12183:2005), which has been technically revised.

Nuclear fuel technology — Controlled-potential coulometric assay of plutonium

1 Scope

This document describes an analytical method for the electrochemical assay of pure plutonium nitrate solutions of nuclear grade, with a total uncertainty not exceeding $\pm 0.2\%$ at the confidence level of 0.95 for a single determination (coverage factor, K = 2). The method is suitable for aqueous solutions containing more than 0.5 g/L plutonium and test samples containing between 4 mg and 15 mg of plutonium. Application of this technique to solutions containing less than 0.5 g/L and test samples containing less than 4 mg of plutonium requires experimental demonstration by the user that applicable data quality objectives will be met.

For some applications, purification of test samples by anion exchange is required before measurement to remove interfering substances when present in significant amounts. Refer to <u>Clause 10</u> for a discussion of interferences and corrective actions. Purification is also appropriate in situations where the purity of the test sample is unknown or when it may fluctuate unpredictably in a manufacturing process.

<u>Clause 11</u> discusses the changes in application of the method and methodology that can be applied and important considerations when selecting measurement parameters, while still remaining within the intended scope of this document.

2 Normative references

There are no normative references in this document.

3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at http://www.electropedia.org/
- ISO Online browsing platform: available at http://www.iso.org/obp

4 Principle

The key steps and their purposes are outlined below:

- test samples are prepared by weighing and then fuming to dryness with sulphuric acid to achieve a
 consistent and stable anhydrous plutonium sulphate salt that is free from chloride, fluoride, nitrate,
 nitrite, hydroxylamine, and volatile organic compounds;
- if needed to remove interferences, dissolve test samples and purify by anion exchange, then fume
 the eluted plutonium solution in the presence of sulphuric acid to obtain the dry plutonium sulphate
 chemical form;
- measure a blank of the nitric acid supporting electrolyte and calculate the background current correction applicable to the electrolysis of the test sample from charging, faradaic, and residual current[1];
- dissolve the dried test sample in the previously measured supporting electrolyte (the blank);