
INTERNATIONAL STANDARD



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Formaldehyde solutions for industrial use — Limit test for inorganic sulphates

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FOREWORD

ISO (the International Organization for Standardization) is a worldwide federation of national standards institutes (ISO Member Bodies). The work of developing International Standards is carried out through ISO Technical Committees. Every Member Body interested in a subject for which a Technical Committee has been set up has the right to be represented on that Committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work.

Draft International Standards adopted by the Technical Committees are circulated to the Member Bodies for approval before their acceptance as International Standards by the ISO Council.

International Standard ISO 2222 was drawn up by Technical Committee ISO/TC 47, *Chemistry*.

It was approved in July 1971 by the Member Bodies of the following countries :

Austria	Ireland	Sweden
Belgium	Israel	Switzerland
Czechoslovakia	Italy	Turkey
Egypt, Arab Rep. of	Netherlands	United Kingdom
France	New Zealand	U.S.A.
Germany	Romania	U.S.S.R.
Hungary	South Africa, Rep. of	

No Member Body expressed disapproval of the document.

Formaldehyde solutions for industrial use — Limit test for inorganic sulphates

WARNING

Formaldehyde is toxic. It is therefore necessary to avoid inhaling its vapour during sampling and testing.

1 SCOPE

This International Standard specifies a limit test for inorganic sulphates in formaldehyde solutions for industrial use.

2 FIELD OF APPLICATION

The method, as described, is applicable when the required limit (x), expressed as a percentage by mass of SO_4 , is not greater than 0,1 % and not less than 0,001 %. If the required limit lies outside that range the mass of the test portion (see 7.1) should be changed and an appropriate adjustment made to the factor $\left(\frac{0,2}{x}\right)$ in 7.3.

3 PRINCIPLE

Comparison of the turbidity obtained by the addition of barium chloride solution to a solution of the sample, in the presence of hydrochloric acid, with that similarly produced from a sulphate solution of known concentration.

4 REAGENTS

Distilled water, or water of equivalent purity, shall be used in the test.

4.1 Hydrochloric acid, approximately N solution.

4.2 Barium chloride ($\text{BaCl}_2 \cdot 2\text{H}_2\text{O}$), 100 g/l solution.

4.3 Sulphate, standard solution 0,1 g of SO_4 per litre.

Place 20,8 ml of exactly 0,1 N sulphuric acid solution in a 1 000 ml one-mark volumetric flask, dilute to the mark and mix.

1 ml of this standard solution contains 0,000 1 g of SO_4 .

5 APPARATUS

Ordinary laboratory apparatus and

5.1 Two matched Nessler cylinders, capacity 100 ml.

6 SAMPLING

Follow the principles given in ISO ...¹⁾.

Attention is drawn to the following recommendation. Place the laboratory sample, representative of the material taken from the bulk, in a clean, dry, and air-tight glass bottle, fitted with a ground glass stopper, of such a size that it is nearly filled by the sample.

If it is necessary to seal this bottle care shall be taken to avoid the risk of contamination.

Owing to polymerization, paraformaldehyde will tend to be deposited on standing and this will occur more rapidly if the temperature is allowed to fall below 25 °C. Accordingly the material shall be sampled as soon as possible after receipt.

7 PROCEDURE

7.1 Test portion and preparation of test solution

Weigh $50 \pm 0,5$ g of the laboratory sample, transfer to a 250 ml one-mark volumetric flask, dilute to the mark and mix.

If the solution is not clear, pass it through a filter paper.

7.2 Preparation of the standard matching solution

To prepare the sulphate solution of known concentration, add to one of the Nessler cylinders (5.1) 4,0 ml of the standard sulphate solution (4.3), dilute to the mark, add 2 ml of the hydrochloric acid solution (4.1) and mix.

7.3 Preparation of the comparison solution

For a material required to contain not more than x % of SO_4 , transfer to the second Nessler cylinder an aliquot, $\frac{0,2}{x}$ ml, of the test solution (7.1). Dilute to the mark, add 2 ml of the hydrochloric acid solution (4.1) and mix.

1) Sampling from the consignment of the product will form the subject of a future International Standard.