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Inorganic fertilizers - Determination of cyanamide nitrogen

EESTI STANDARDI EESSÕNA

NATIONAL FOREWORD

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ICS 65.080

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EUROPEAN STANDARD

EN 15562

NORME EUROPÉENNE

EUROPÄISCHE NORM

August 2023

ICS 65.080

Supersedes EN 15562:2009

English Version

Inorganic fertilizers - Determination of cyanamide nitrogen

Engrais inorganiques - Dosage de l'azote cyanamidé

Anorganische Düngemittel - Bestimmung des Cyanamidstickstoffs

This European Standard was approved by CEN on 26 June 2023.

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This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the CEN-CENELEC Management Centre has the same status as the official versions.

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EUROPEAN COMMITTEE FOR STANDARDIZATION
COMITÉ EUROPÉEN DE NORMALISATION
EUROPÄISCHES KOMITEE FÜR NORMUNG

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Contents		Page
European foreword		3
1	Scope	4
2	Normative references	4
3	Terms and definitions	4
4	Principle	4
5	Reagents	4
6	Apparatus and equipment	5
7	Sampling and sample preparation	5
8	Procedure	6
8.1	Preparation of the solution for analysis	6
8.2	Analysis of the solution	6
8.3	Blank test	7
8.4	Control test	7
9	Expression of the result	7
10	Test report	8
Bibliography		9

European foreword

This document (EN 15562:2023) has been prepared by Technical Committee CEN/TC 260 “Fertilizers and liming materials”, the secretariat of which is held by DIN.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by February 2024, and conflicting national standards shall be withdrawn at the latest by February 2024.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN shall not be held responsible for identifying any or all such patent rights.

This document supersedes EN 15562:2009.

In comparison with the previous edition EN 15562:2009, the following technical modifications have been made:

- technical changes have been made to the description of the procedure (Clause 8) and to Table 1 (8.2);
- deletion of distillation apparatus drawings.

This document has been prepared under a standardization request addressed to CEN by the European Commission. The Standing Committee of the EFTA States subsequently approves these requests for its Member States.

Any feedback and questions on this document should be directed to the users’ national standards body. A complete listing of these bodies can be found on the CEN website.

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1 Scope

This document specifies a method for the determination of cyanamide nitrogen in fertilizers. The method is applicable to calcium cyanamide and mixtures of calcium cyanamide with calcium nitrate.

2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 1482-2, *Fertilizers and liming materials — Sampling and sample preparation — Part 2: Sample preparation*

EN 12944-1, *Fertilizers and liming materials — Vocabulary — Part 1: General terms*

EN 12944-2, *Fertilizers and liming materials — Vocabulary — Part 2: Terms relating to fertilizers*

3 Terms and definitions

For the purposes of this document, the terms and definitions given in EN 12944-1 and EN 12944-2 apply. ISO and IEC maintain terminology databases for use in standardization at the following addresses:

- IEC Electropedia: available at <https://www.electropedia.org/>
- ISO Online browsing platform: available at <https://www.iso.org/obp>

4 Principle

Precipitation of cyanamide nitrogen as a silver complex. Digestion in sulfuric acid with the aid of a catalyst. Distillation of the ammonia from an alkaline solution, absorption in an excess of sulfuric acid volumetric solution and back-titration with sodium or potassium hydroxide volumetric solution.

5 Reagents

Use only reagents of recognized analytical grade and water with an electrical conductivity $< 0,5$ mS/m (at 25 °C).

5.1 Glacial acetic acid

5.2 Ammonia solution, containing 10 % of ammonia gas by mass, $\rho_{20} = 0,96$ g/ml

5.3 Ammoniacal silver solution, according to Tollens

Mix 500 ml of 10 % silver nitrate (AgNO_3) solution in water with 500 ml of 10 % ammonia solution (5.2).

Do not expose unnecessarily to light, heat or air. The solution normally keeps for years. As long as the solution remains clear, the reagent is of good quality.

5.4 Sulfuric acid, mass concentration $\rho_{20} = 1,84$ g/ml

5.5 Potassium sulphate, analytical grade