
INTERNATIONAL STANDARD



993

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Potassium hydroxide for industrial use – Determination of sulphate content – Barium sulphate gravimetric method

Hydroxyde de potassium à usage industriel – Dosage des sulfates – Méthode gravimétrique à l'état de sulfate de baryum

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FOREWORD

ISO (the International Organization for Standardization) is a worldwide federation of national standards institutes (ISO Member Bodies). The work of developing International Standards is carried out through ISO Technical Committees. Every Member Body interested in a subject for which a Technical Committee has been set up has the right to be represented on that Committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work.

Draft International Standards adopted by the Technical Committees are circulated to the Member Bodies for approval before their acceptance as International Standards by the ISO Council.

Prior to 1972, the results of the work of the Technical Committees were published as ISO Recommendations; these documents are now in the process of being transformed into International Standards. As part of this process, Technical Committee ISO/TC 47 has reviewed ISO Recommendation R 993 and found it technically suitable for transformation. International Standard ISO 993 therefore replaces ISO Recommendation R 993-1969 to which it is technically identical.

ISO Recommendation R 993 was approved by the Member Bodies of the following countries :

Austria	India	Romania
Belgium	Ireland	South Africa, Rep. of
Brazil	Israel	Spain
Chile	Japan	Switzerland
Cuba	Korea, Dem. P. Rep. of	Thailand
Czechoslovakia	Netherlands	Turkey
Egypt, Arab Rep. of	New Zealand	United Kingdom
Germany	Poland	U.S.S.R.
Hungary	Portugal	Yugoslavia

The Member Bodies of the following countries expressed disapproval of the Recommendation on technical grounds :

France*
Italy
U.S.A.

* Subsequently, this Member Body approved the Recommendation.

The Member Body of the following country disapproved the transformation of ISO/R 993 into an International Standard :

United Kingdom

Potassium hydroxide for industrial use – Determination of sulphate content – Barium sulphate gravimetric method

1 SCOPE AND FIELD OF APPLICATION

This International Standard specifies a barium sulphate gravimetric method for the determination of the sulphate content of potassium hydroxide for industrial use.

The method is applicable to products having a sulphate content, expressed as potassium sulphate and calculated on KOH, equal to or greater than 0,10 % (*m/m*).

2 REFERENCE

ISO 2466, *Potassium hydroxide for industrial use – Sampling – Test sample – Preparation of the main solution for carrying out certain determinations.*

3 PRINCIPLE

Precipitation of sulphate as barium sulphate in dilute hydrochloric acid.

Separation of the precipitate, heating at 800 ± 25 °C and weighing.

4 REAGENTS

During the analysis, use only reagents of recognized analytical grade and only distilled water or water of equivalent purity.

4.1 Hydrochloric acid, ρ approximately 1,19 g/ml, about 38 % (*m/m*) or 12 N solution.

4.2 Sulphuric acid, ρ approximately 1,84 g/ml, about 96 % (*m/m*) or 36 N solution.

4.3 Barium chloride dihydrate ($\text{BaCl}_2 \cdot 2\text{H}_2\text{O}$), 122 g/l or approximately 1 N solution.

4.4 Silver nitrate, 5 g/l nitric solution.

Dissolve 0,5 g of silver nitrate in a small amount of water, add 10 ml of nitric acid solution, ρ approximately 1,40 g/ml, and dilute to 100 ml.

4.5 Methyl orange, 0,5 g/l solution.

5 APPARATUS

Ordinary laboratory apparatus and

5.1 Platinum crucible, top diameter approximately 30 mm and depth approximately 30 mm.

5.2 Electric oven, capable of being controlled at 110 ± 2 °C.

5.3 Electric furnace, capable of being controlled at 800 ± 25 °C.

6 PROCEDURE

6.1 Test portion

Weigh, to the nearest 0,01 g, a mass of the solid or liquid test sample corresponding to approximately 10 g of KOH (see ISO 2466).

6.2 Preparation of the test solution

Place the test portion (6.1) in a 600 ml beaker. In the case of solid material, dissolve the test portion in about 100 ml of water; in the case of liquid material, dilute to approximately 100 ml. Add 5 drops of the methyl orange solution (4.5) and slowly, while stirring, the volume of the hydrochloric acid solution (4.1) required for neutralization. Then add *immediately* 2 ml in excess of this acid.

Transfer the solution quantitatively to a 200 ml one-mark volumetric flask, dilute to the mark and mix.

Filter on a dry, slow-speed, ashless filter paper of diameter about 90 mm, and discard the first 10 ml of the filtrate.

6.3 Determination

Place 100,0 ml of the test solution (6.2) in a beaker of suitable capacity (for example 600 ml). Bring to the boil, stir continuously and add 10 ml of the barium chloride solution (4.3) drop by drop (the addition should take about 90 s).

Maintain boiling for 2 min, stirring all the time. Heat on a boiling water bath for 2 h; stop heating and allow to stand for about 16 h.