
INTERNATIONAL STANDARD



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INTERNATIONAL ORGANIZATION FOR STANDARDIZATION • МЕЖДУНАРОДНАЯ ОРГАНИЗАЦИЯ ПО СТАНДАРТИЗАЦИИ • ORGANISATION INTERNATIONALE DE NORMALISATION

Phosphoric acid for industrial use — Determination of arsenic content — Silver diethyldithiocarbamate photometric method

Acide phosphorique à usage industriel — Dosage de l'arsenic — Méthode photométrique au diéthylthiocarbamate d'argent

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FOREWORD

ISO (the International Organization for Standardization) is a worldwide federation of national standards institutes (ISO Member Bodies). The work of developing International Standards is carried out through ISO Technical Committees. Every Member Body interested in a subject for which a Technical Committee has been set up has the right to be represented on that Committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work.

Draft International Standards adopted by the Technical Committees are circulated to the Member Bodies for approval before their acceptance as International Standards by the ISO Council.

International Standard ISO 3359, was drawn up by Technical Committee ISO/TC 47, *Chemistry*, and circulated to the Member Bodies in April 1974.

It has been approved by the Member Bodies of the following countries :

Austria	India	South Africa, Rep. of
Belgium	Israel	Spain
Bulgaria	Italy	Switzerland
Chile	Netherlands	Thailand
Czechoslovakia	New Zealand	Turkey
France	Poland	United Kingdom
Germany	Portugal	Yugoslavia
Hungary	Romania	

No Member Body expressed disapproval of the document.



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AMENDMENT

Foreword (Inside front cover)

The ISO Member Body for the Arab Republic of Egypt has now approved this International Standard. The Arab Republic of Egypt should therefore be included in the list of countries whose Member Bodies have approved the document.

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Phosphoric acid for industrial use – Determination of arsenic content – Silver diethyldithiocarbamate photometric method

1 SCOPE AND FIELD OF APPLICATION

This International Standard specifies a silver diethyldithiocarbamate photometric method for the determination of the arsenic content of phosphoric acid for industrial use.

The method can be used for the determination of quantities of arsenic, expressed as As, greater than 0,1 mg/kg.

2 REFERENCES

ISO 2590, *General method for the determination of arsenic – Silver diethyldithiocarbamate photometric method.*

ISO 4285, *Phosphoric acid for industrial use – Sampling technique.*¹⁾

3 PRINCIPLE

See ISO 2590, clause 3.

4 REAGENTS

The reagents listed in clause 4 of ISO 2590 and

4.9 Bromide-bromate solution, approximately 2 N.

Dissolve 20 g of potassium bromide (KBr) and 5,20 g of potassium bromate (KBrO₃) in water and dilute to 100 ml.

5 APPARATUS

See ISO 2590, clause 5.

6 PROCEDURE

WARNING – See ISO 2590, clause 6.

6.1 Test portion and preparation of the test solution

Weigh, to the nearest 1 mg, into the conical flask (5.1.1) a quantity of the test sample containing between 1 and 20 µg of As. This quantity shall not be greater than 10 g. Dilute with 10 ml of water, add 1 ml of the bromide-bromate

solution (4.9), and warm the solution on a boiling water bath. Then add to the hot solution, in small portions, the bromide-bromate solution until the yellow colour from the released bromine persists for about 5 min. Add 1 ml of the bromide-bromate solution in excess and continue heating on the boiling water bath until the disappearance of the yellow bromine colour. Cool the solution to room temperature, dilute to about 30 ml and add 10 ml of the hydrochloric acid solution (4.1).

NOTE – If hydrochloric acid solution of concentration approximately 12 N is not available, a solution of lower concentration may be used, by adding the corresponding stoichiometric quantity of HCl and adjusting the dilution to reach the same final volume of 40 ml.

6.2 Blank test

See ISO 2590, sub-clause 6.2.

6.3 Preparation of the calibration curve

See ISO 2590, sub-clause 6.3.

6.4 Determination

Follow the procedure specified in sub-clause 6.4 of ISO 2590.

7 EXPRESSION OF RESULTS

By means of the calibration curve (see 6.3.3), determine the quantity of arsenic (As) corresponding to the value of the photometric measurement of the test solution.

The arsenic content, expressed in milligrams of arsenic (As) per kilogram, is given by the formula

$$\frac{m_1}{m_0}$$

where

m_0 is the mass, in grams, of the test portion (6.1);

m_1 is the mass, in micrograms, of As found in the test solution.