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**Photography — Processing wastes —  
Analysis of cyanides —**

**Part 1:**

Determination of hexacyanoferrate(II) and  
hexacyanoferrate(III) by spectrometry

*Photographie — Effluents de traitement — Analyse des cyanures —*

*Partie 1: Détermination de l'hexacyanoferrate(II) et de  
l'hexacyanoferrate(III) par spectrométrie*

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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 7766-1 was prepared by Technical Committee ISO/TC 42, *Photography*.

ISO 7766 consists of the following parts, under the general title *Photography — Processing wastes — Analysis of cyanides*:

- *Part 1: Determination of hexacyanoferrate(II) and hexacyanoferrate(III) by spectrometry*

Further parts are in preparation.

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## Introduction

This part of ISO 7766 is one of a series devoted to the analysis of photographic wastes; it encompasses the field of analysis of complexed iron cyanides (hexacyanoferrates) in photographic effluents.

Some of the chemicals specified in the test procedures are caustic, toxic, or otherwise hazardous. Specific warning, caution and danger notices are noted but, in addition, the normal precautions required during the performance of any chemical procedure should be exercised at all times.

In the case of effluents, the photographic laboratory can best establish conformity to regulations by appropriate chemical analysis. In some cases, in-house analyses will be possible; often the use of an outside laboratory will be required.

Complexed cyanides are used in the bleaching stage of colour photographic processing and it is, therefore, possible for the cyanide portion of the resulting photographic effluent to reach a reportable level. Complexed cyanides contribute to the total cyanides in those tests in which the sample preparation breaks down the complexed cyanides. It is the purpose of this part of ISO 7766 to provide a method giving an independent estimate of the cyanide present as complexes. Due to the chemical behaviour of cyanide complexes, it is not possible to specify a single method for the quantitative determination of complex cyanides in these effluents. This part of ISO 7766 provides an analytical procedure for the determination of the hexacyanoferrate complex.

The analysis of cyanide is covered in various aspects in the following International Standards:

- ISO 6703-1:1984, *Water quality — Determination of cyanide — Part 1: Determination of total cyanide*;
- ISO 6703-2:1984, *Water quality — Determination of cyanide — Part 2: Determination of easily liberatable cyanide*;
- ISO 6703-4:1985, *Water quality — Determination of cyanide — Part 4: Determination of cyanide by diffusion at pH 6*.

### NOTES

1 Easily liberatable cyanides are defined as substances with cyanide groups and a measurable hydrocyanic acid vapour pressure at pH 4 and room temperature.

2 Cyanide diffuses as hydrogen cyanide (HCN) at room temperature from a solution at pH 6; the procedure determines cyanide from simple compounds of cyanide and easily dissociated complexes.

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# Photography — Processing wastes — Analysis of cyanides —

## Part 1:

### Determination of hexacyanoferrate(II) and hexacyanoferrate(III) by spectrometry

**WARNING — Reagents and samples specified in this part of ISO 7766 contain cyanide. Although the cyanide is in a complexed form, it is possible for toxic hydrogen cyanide gas to form if the reagents are not treated correctly. Use these materials in a well-ventilated fume hood.**

## 1 Scope

This part of ISO 7766 establishes a test method for the determination of hexacyanoferrate(II) (ferrocyanide) and hexacyanoferrate(III) (ferricyanide), referred to hereafter as  $\text{Fe}(\text{CN})_6$ , in photographic processing effluents.<sup>1)</sup> Results are reported as hexacyanoferrate,  $\text{Fe}(\text{CN})_6$ .

## 2 Normative references

The following standards contain provisions which, through reference in this text, constitute provisions of this part of ISO 7766. At the time of publication, the editions indicated were valid. All standards are subject to revision, and parties to agreements based on this part of ISO 7766 are encouraged to investigate the possibility of applying the most recent editions of the standards indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 385-1:1984, *Laboratory glassware — Burettes — Part 1: General requirements.*

ISO 385-2:1984, *Laboratory glassware — Burettes — Part 2: Burettes for which no waiting time is specified.*

ISO 648:1977, *Laboratory glassware — One-mark pipettes.*

ISO 835-1:1981, *Laboratory glassware — Graduated pipettes — Part 1: General requirements.*

ISO 835-2:1981, *Laboratory glassware — Graduated pipettes — Part 2: Pipettes for which no waiting time is specified.*

ISO 835-3:1981, *Laboratory glassware — Graduated pipettes — Part 3: Pipettes for which a waiting time of 15 s is specified.*

ISO 835-4:1981, *Laboratory glassware — Graduated pipettes — Part 4: Blow-out pipettes.*

ISO 1042:1983, *Laboratory glassware — One-mark volumetric flasks.*

ISO 3696:1987, *Water for analytical laboratory use — Specification and test methods.*

1) Significant levels of thiosulfate (hypo) can result in interference. Significant levels of thiosulfate in processing-machine effluents result in oxidation of the thiosulfate by iron(III). Sulfur will form, increasing the absorbance of the test. Also, thiosulfate may deplete the added iron(III) leaving nothing for the colour reaction. The method is applicable to effluents from buildings where the thiosulfate would be diluted by the rest of the building wastes and this would eliminate the interference.

ISO 4788:1980, *Laboratory glassware — Graduated measuring cylinders.*

ISO 5667-1:1980, *Water quality — Sampling — Part 1: Guidance on the design of sampling programmes.*

ISO 5667-2:1991, *Water quality — Sampling — Part 2: Guidance on sampling techniques.*

ISO 5667-3:1985, *Water quality — Sampling — Part 3: Guidance on the preservation and handling of samples.*

ISO 5725:1986, *Precision of test methods — Determination of repeatability and reproducibility for a standard test method by inter-laboratory tests.*

ISO 6703-1:1984, *Water quality — Determination of cyanide — Part 1: Determination of total cyanide.*

ISO 6703-2:1984, *Water quality — Determination of cyanide — Part 2: Determination of easily liberatable cyanide.*

ISO 6703-4:1985, *Water quality — Determination of cyanide — Part 4: Determination of cyanide by diffusion at pH 6.*

### 3 Safety and hazards

#### 3.1 Hazard warnings

Some of the chemicals specified in the test procedures are caustic, toxic, or otherwise hazardous. Safe laboratory practice for the handling of chemicals requires the use of safety glasses or goggles, rubber gloves and other protective apparel such as face masks or aprons where appropriate. Specific danger notices are given in the text and footnotes for particularly dangerous materials, but normal precautions are required during the performance of any chemical procedure at all times. The first time that a hazardous material is noted in the test procedure section, the hazard will be indicated by the word "DANGER" followed by a symbol consisting of angle brackets "<>" containing a letter which designates the specific hazard. A double bracket "<<>>" will be used for particularly perilous situations. In subsequent statements involving handling of these hazardous materials, only the hazard symbol consisting of the brackets and letter(s) will be displayed. Furthermore, for a given material, the hazard symbol will be used only once in a single paragraph.

Detailed warnings for handling chemicals and their diluted solutions are beyond the scope of this part of ISO 7766.

**Employers shall provide training and health and safety information in conformance with legal requirements.**

The hazard symbol system used in this part of ISO 7766 is intended to provide information to the users and is not meant for compliance with any legal requirements for labelling as these vary from country to country.

**It is strongly recommended that anyone using these chemicals obtain from the manufacturer pertinent information about the hazards, handling, use and disposal of these chemicals.**

#### 3.2 Hazard information code system

- < B > Harmful if inhaled. Avoid breathing dust, vapour, mist or gas. Use only with adequate ventilation.
- < C > Harmful if contact occurs. Avoid contact with eyes, skin or clothing. Wash thoroughly after handling.
- < S > Harmful if swallowed. Wash thoroughly after handling. If swallowed, obtain medical attention immediately.
- << S >> May be fatal if swallowed. If swallowed, obtain medical attention immediately.
- < F > Will burn. Keep away from heat, sparks and open flame. Use with adequate ventilation.<sup>2)</sup>
- < O > Oxidizer. Contact with other material may cause fire. Do not store near combustible materials.

#### 3.3 Safety precautions

**ALL PIPETTE OPERATIONS SHALL BE PERFORMED WITH A PIPETTE BULB OR PLUNGER PIPETTE. Failure to observe this warning notice can result in cyanide poisoning. THIS IS A CRITICAL SAFETY WARNING!**

**Digestion procedures shall be performed in a fume hood. Hydrogen cyanide or other toxic substances may be evolved.**

**Safety glasses shall be worn for all laboratory work.**

### 4 Principle

A sample of effluent, or diluted effluent, is treated with a mixture of iron(II) and iron(III) ions. If  $\text{Fe}(\text{CN})_6$  is present, a blue suspension will form. In the range

2) The flammable warning symbol < F > will not be used for quantities of common solvents under 1 litre.